

Lecture 1 Review : Fluid Properties

Density ρ = mass / unit vol

Pressure P = force / unit area \perp to any surface

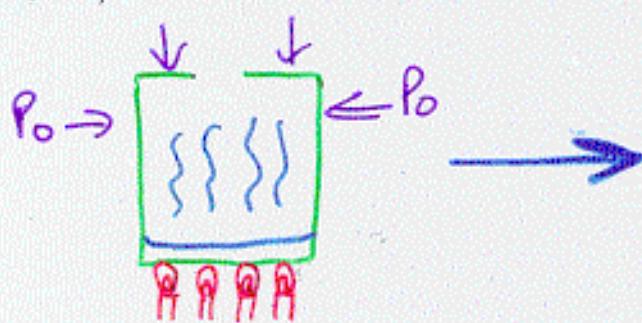
Hydrostatic Pressure gradient $\frac{dP}{dy} = -\rho g$ [Pa/m]

For many liquids $\rho \approx \text{constant}$ ("incompressible")

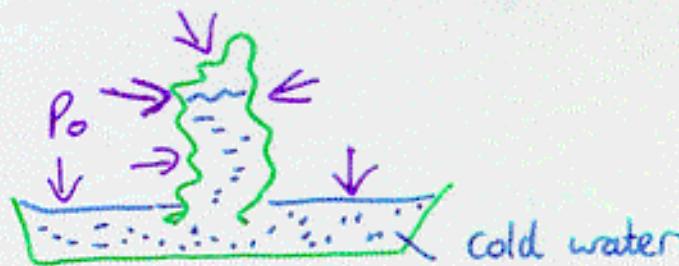
$$\Rightarrow P_1 - P_2 = \rho g (y_2 - y_1) \text{ or } \Delta P = \rho g \underline{h} \quad \begin{matrix} \text{depth} \\ \text{only} \end{matrix}$$

For gases, ρ changes with Pressure (and Temperature)

e.g. "Crushing Can" demo:



heat: steam at
 $P = 1 \text{ atm}$



rapid cooling reduces steam
pressure inside can

Results: • Can crushed by outside air pressure

• air pressure forces water from bath into can.

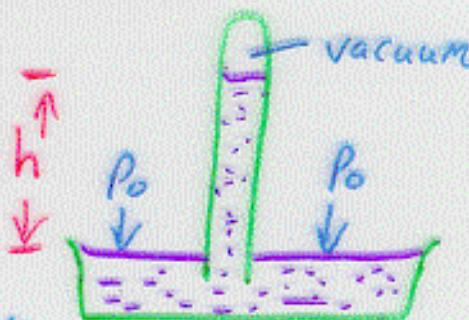
Using Fluids to Measure Pressure

1. Mercury (Hg) barometer:

atm. pressure supports

fluid column above level of bath

$$\Rightarrow P_0 = \rho g h$$



Given standard values of ρ and g , can measure P_0

in terms of height h . $1 \text{ torr} = 1 \text{ mm Hg}$

$$1 \text{ atm} \approx 760 \text{ torr} \approx 10^5 \text{ Pa}$$

Q: Why not use water?

For $1 \text{ atm} \approx 10^5 \text{ Pa}$, $\rho = 10^3 \text{ kg/m}^3$, height of column $h = \frac{P_0}{\rho g} = 10 \text{ m}$ or 33 ft !

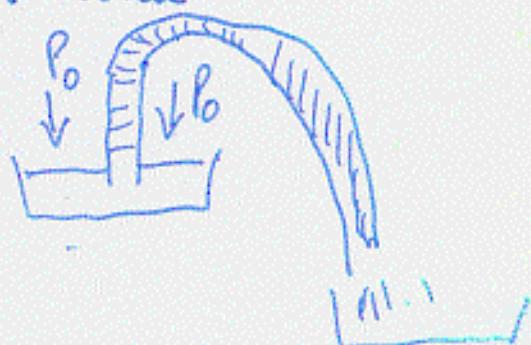
Also, water vapor destroys vacuum at top



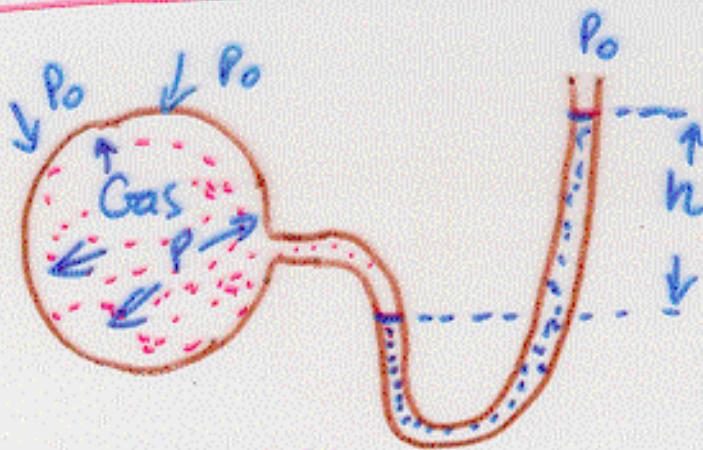
BUT: Can use atm. pressure to "lift" water

up over a barrier $\approx 10 \text{ m}$ high

i.e. a siphon



2. Open-tube Manometer: Pressure differences



Pressure difference between gas and atmosphere

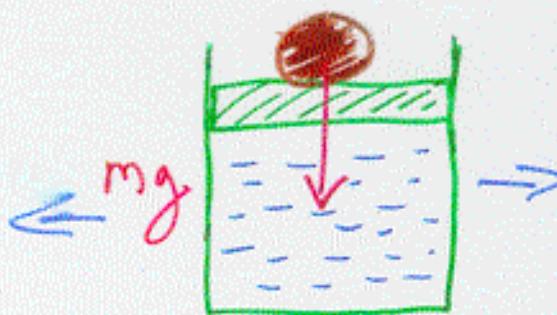
$$P - P_0 = \rho g h$$

- can measure h as a warning that gas vessel (aircraft cabin) will explode if $(P - P_0)$ gets too high

Pascal's Principle

- Already assumed / demo'd

"Change in pressure applied to incompressible fluid
 transmitted to every portion of the fluid"
 (transmitted at speed of sound - fast!)



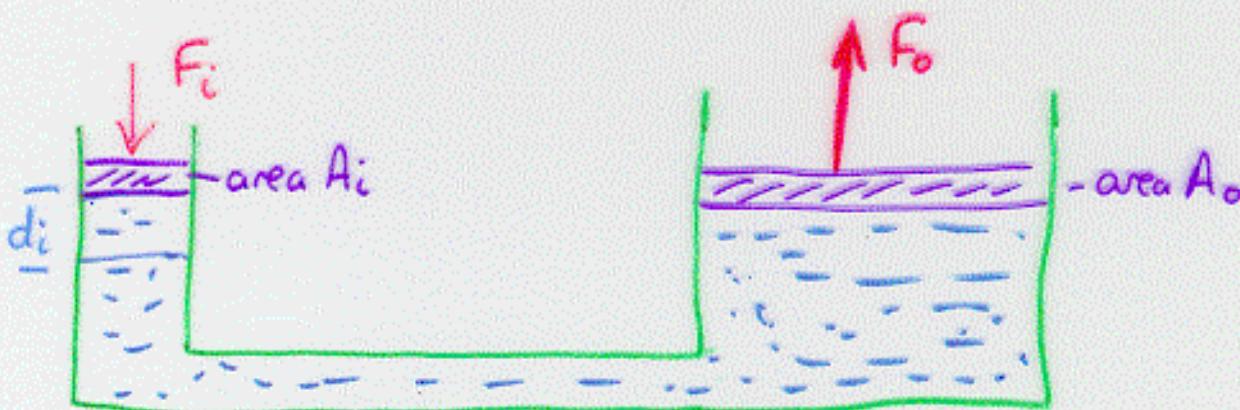
e.g. add weight to top of fluid
 → transmitted to bottom and
 side walls

(Remember: at any point in fluid, pressure supports
 weight of everything above it)

$$\text{At top of fluid: } P = P_0 + \frac{mg}{A}$$

$$\text{At depth } h \quad P = P_0 + \frac{mg}{A} + \rho gh$$

Hydraulic Lever



Apply input force \$F_i \Rightarrow\$ increases \$P\$ throughout fluid
by \$\Delta P = F_i/A_i\$ (Pascal)

\$\Rightarrow\$ net force on larger piston \$F_o = \Delta P/A_o

$$\text{i.e. } F_o = \frac{A_o}{A_i} F_i \quad : \text{force magnified!}$$

But: no free lunch. If \$F_i\$ moves piston down by \$d_i\$

$$\text{input work} = F_i d_i$$

$$\text{volume displaced} = A_i d_i = V$$

$$\therefore \text{output piston moves up by } d_o = \frac{V}{A_o} = \frac{A_i d_i}{A_o}$$

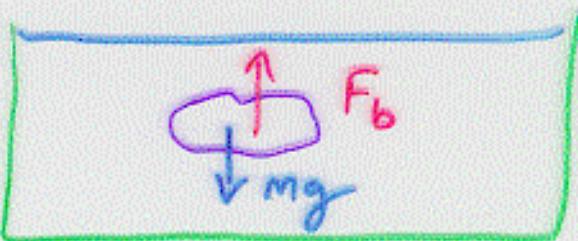
$$\Rightarrow \text{output work } F_o d_o = \frac{A_o}{A_i} F_i \cdot \frac{A_i}{A_o} d_i$$

$$\text{i.e. } F_o d_o = F_i d_i$$

\$\therefore\$ Work (energy) conserved

Archimedes Principle - Eureka!

Buoyant Force



Isolate a piece of fluid, any shape or size

At eqm., weight of fluid piece is supported by pressure diff. above + below:

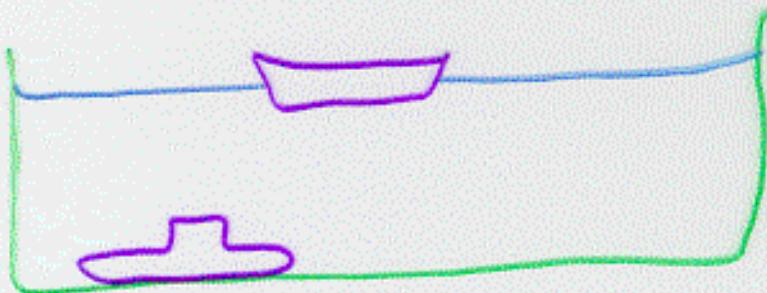
$$\text{i.e. a } \underline{\text{buoyant force}} \quad F_b = \int P.dA = m g = \underline{\rho V g}$$

Now "remove" fluid piece and replace with something else, mass M

Buoyant force F_b still provided by fluid above / below

\Rightarrow "A body immersed in a fluid experiences a buoyant force F_b = weight of the fluid displaced".
(occupying same volume)

Archimedes cont/d.



"Effective weight" of object (downwards)

$$= Mg - \rho Vg$$

- Can be used to simulate reduced gravity
(bottom of pool = moon)

Or: if $M < \rho V$, i.e. $\rho_{obj} = \frac{M}{V} < \rho$,

object is pushed upwards until it floats,
displacing a mass = M of fluid

e.g. ships with steel hulls

Subs, fishes, whales control F_b by changing their
average density relative to water

i.e. $\frac{M}{V}$