

Physics 2C: Fluids, Thermodynamics, Optics and Waves

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Additions to Syllabus:

Prof. office hours: Wed 2-3pm, 312 SERF bldg.

Tuesday lecture: start time 8-8:10am, HSS22
(babysitter)

Muir campus

Things you need to know

Homeworks: • Check the WWW page for assignments.
(0%) • Solutions posted ~ Wednesday's lecture
• Do the problems yourself!

Quizzes: • every Friday, incl. this Friday
(65%) • best 5 count towards grade, no make-ups

Final Exam: • Tues March 20, 11:30am
(35%) • inform me of conflicts asap!

"Rules"

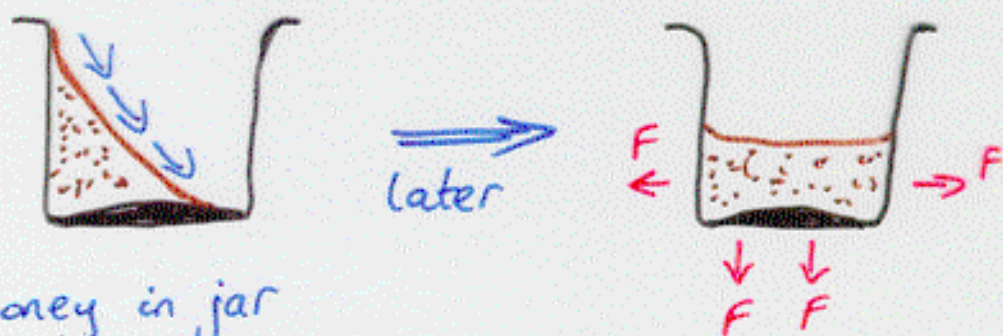
- All tests "closed book"
 - bring new, empty blue book to each quiz
- Lectures define course content, not textbook
- No, I don't "grade on a curve".

FLUIDS : stuff (matter) that flows!
(liquids and gases)

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i.e. cannot sustain tangential (shear) forces

e.g.



honey in jar

- time taken $\propto$ viscosity
- conforms to container's shape
- - fluids do exert force $\perp$ to boundaries (F)
- fluids "seek their own level" at equilibrium

Physical Properties

1. Density $\rho \equiv \text{mass} / \text{unit volume}$ [kg / m^3]

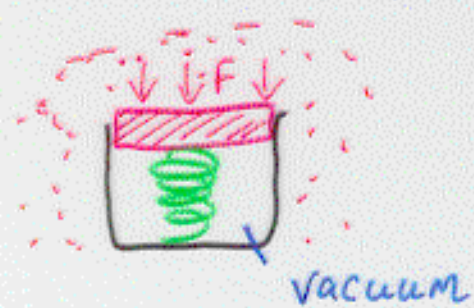
e.g. water has $\rho = 1000 \text{ kg} / \text{m}^3$

air in room $\approx 1.2 \text{ kg} / \text{m}^3$

Note: pump air into tire $\Rightarrow$ tire gets heavier

$$\text{Weight of air } mg = \rho V g$$

2. Pressure $P = \text{force} / \text{unit area}$



gas pressure on piston
compresses spring
 $P = F/A$

(Kinetic theory of gases will explain origin of force F).

Units: $1 \text{ N/m}^2 \equiv 1 \text{ Pascal (Pa)}$

$1 \text{ atm} \approx 10^5 \text{ Pa} \approx 760 \text{ torr} \approx 15 \text{ psi}$ (Appendix D)
mm Hg

Note:

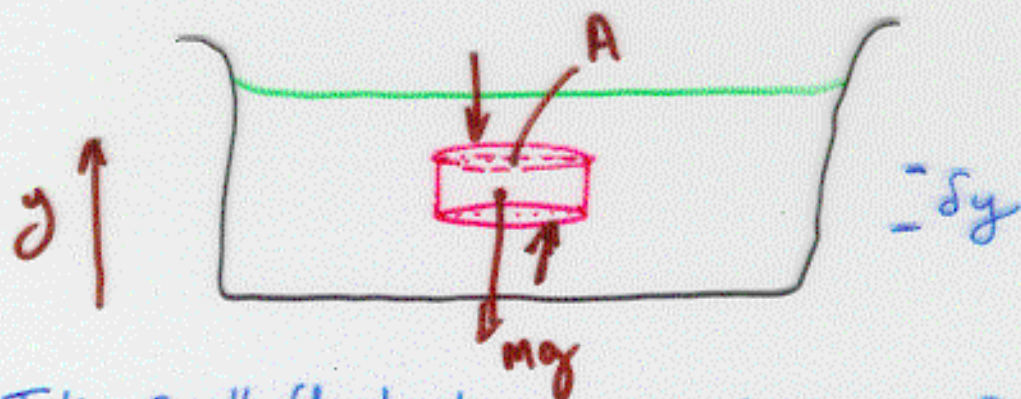
Gases - density ρ changes with pressure (and temperature)

e.g. can "squeeze" more air into constant-volume tire

Liquids - density $\sim$ constant with pressure
"incompressible"

Fluids at Rest - Hydrostatic Pressure

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Take small fluid element, area A , height δy

At eqm., weight of element supported by pressure difference.

$$\text{Force on bottom of element} = P \cdot A$$

top

$$= (P + \delta P) \cdot A$$

$$\text{So } P \cdot A = (P + \delta P) \cdot A + \underbrace{\rho A \delta y}_{\text{volume}} g$$

= weight

$$\text{i.e. } \delta P = -\rho g \delta y$$

$$\Rightarrow \frac{dP}{dy} = -\rho g \quad (\text{pressure gradient})$$

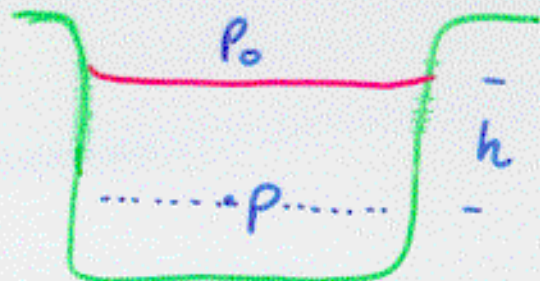
$$\text{OR } P_1 - P_2 = - \int_{y_1}^{y_2} \rho g \cdot dy$$

Pressure at any level y = weight of all the stuff above that level (at eqm.)

$$P_1 - P_2 = - \int \rho g dy$$

e.g. for incompressible liquids, $\rho = \text{constant}$

$$\Rightarrow \underline{P = P_0 + \rho g h \text{ at depth } h :}$$



P_0 = atm. pressure at surface = weight of air column

Example: For a human 2 m tall, show that the blood pressure difference between head and feet is approx. 2×10^4 Pa. Give 3 significant figures

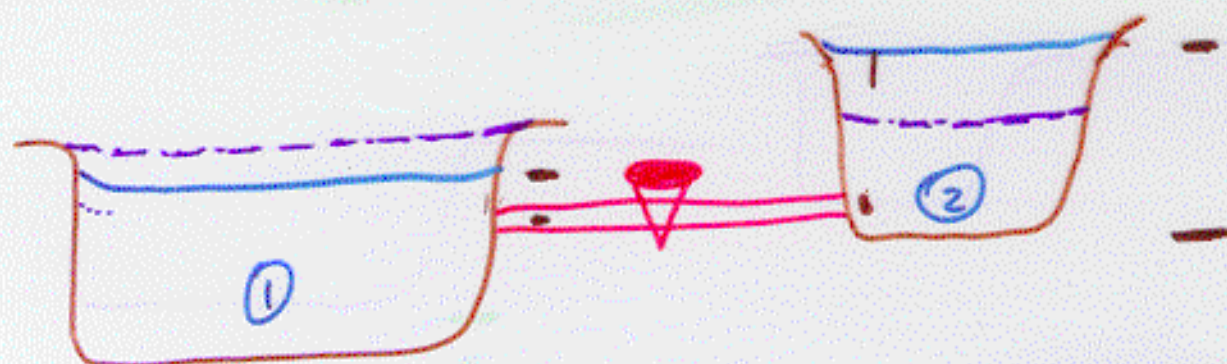
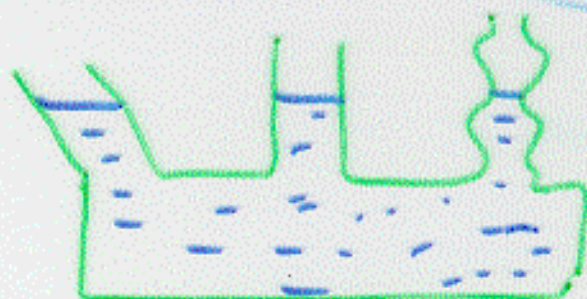
$$\rho_{\text{blood}} = 1.1 \times 10^3 \text{ kg/m}^3$$

$$g = 9.81 \text{ m/s}^2$$

"Fluids seek their own level"

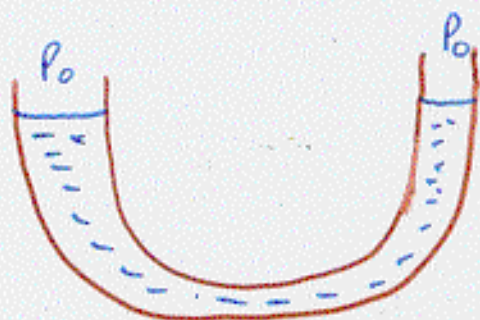
Reason: Pressure depends only on depth

e.g.



Open valve in pipe: higher pressure in ② forces liquid from ② to ① until each end of pipe is at same pressure, i.e. same depth

c.f. "U-tube"



level does not depend on width of tube.