

Answers to Questions

5. $B > A = C$

12. $H = kA \frac{T_H - T_C}{L}$

$\Rightarrow A = B = C > D = 0$

13. at the temperature of your finger.

Exercises & Problems

4P. (T_L, P_L) = (temperature, pressure) of the left-hand thermometer

(T_R, P_R) = (temp., pressure) of the right-hand thermometer

$\Rightarrow \frac{T_L}{273.16 \text{ K}} = \frac{P_L}{P_3} \quad \& \quad \frac{T_R}{273.16 \text{ K}} = \frac{P_R}{P_3}$

where P_3 = pressure of the gas at the triple point of water (= 273.16 K)

Subtract the second eqn. from the first.

$T_L - T_R = (273.16 \text{ K}) \cdot \left(\frac{P_L - P_R}{P_3} \right)$

Take $T_L = 373.125 \text{ K}$ (the boiling point of water) and
 $T_R = 273.16 \text{ K}$ (the triple point of water)
 then $P_L - P_R = 120 \text{ mm of mercury}$.

$\Rightarrow 373.125 \text{ K} - 273.16 \text{ K} = (273.16 \text{ K}) \left(\frac{120 \text{ mm}}{P_3} \right)$

$\therefore P_3 = 328 \text{ mm}$

(2)

Now let $T_L = 273.16 \text{ K}$ and $T_R = \text{unknown}$.
 then $P_L - P_R = 90 \text{ mm}$

$$\Rightarrow 273.16 \text{ K} - T_R = (273.16 \text{ K}) \left(\frac{90 \text{ mm}}{328 \text{ mm}} \right)$$

$$\therefore T_R = 348 \text{ K}$$

26E. $\Delta V = \beta V \Delta T$ and $\beta = 3\alpha$

For aluminum $\beta = 3 \times (23 \times 10^{-6} / ^\circ\text{C})$
 $= 69 \times 10^{-6} / ^\circ\text{C}$

$$\Delta V = \Delta V_{\text{glycerin}} - \Delta V_{\text{aluminum}}$$

$$= \beta_{\text{gly}} V \Delta T - \beta_{\text{alum}} V \Delta T$$

$$\left(\begin{array}{l} V = 100 \text{ cm}^3 \\ \Delta T = \text{6}^\circ\text{C} \end{array} \right)$$

$$= (\beta_{\text{gly}} - \beta_{\text{alum}}) V \Delta T$$

$$= (5.1 \times 10^{-4} / ^\circ\text{C} - 0.69 \times 10^{-4} / ^\circ\text{C}) (100 \text{ cm}^3) (6^\circ\text{C})$$

$$= 26.5 \times 10^{-2} \text{ cm}^3$$

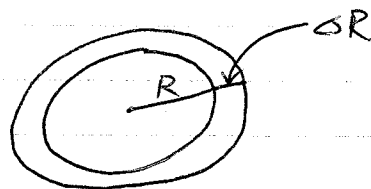
$$= 0.26 \text{ cm}^3$$

will spill out of the cup.

(5)

34P. (a) D : diameter

$$\frac{\Delta D}{D} = 0.18 \times 10^{-2}$$



$$\frac{\Delta A}{A} = \frac{2\pi R \cdot \Delta R}{\pi R^2} = 2 \cdot \frac{\Delta R}{R} = 2 \cdot \frac{\Delta D/2}{D/2} = 2 \cdot \frac{\Delta D}{D}$$

$$\therefore \frac{\Delta A}{A} = 2 \cdot \frac{\Delta D}{D} = 0.36 \times 10^{-2}$$

0.36% area increase.

(b) $\Delta L = \alpha L \Delta T$ applies to every dimension of the object. ~~NOT~~

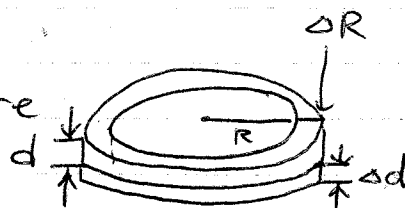
Therefore, thickness increases by 0.18%.

(c) You might notice that from (a), volume increases by $3 \times \frac{\Delta D}{D}$.

This is true actually.

$$\frac{\Delta V}{V} = 3 \times \frac{\Delta D}{D} = 0.54 (\%)$$

To prove this, consider the figure



$$\Delta V = \pi(R + \Delta R)^2 \cdot (d + \Delta d) - \pi R^2 d$$

$$= \pi \left(R^2 \Delta d + \underbrace{2R \Delta R d}_{2^{\text{nd}}} + \underbrace{2R \Delta R \Delta d}_{2^{\text{nd}}} + \underbrace{\Delta R^2 d}_{2^{\text{nd}}} + \Delta R^2 \Delta d \right)$$

Ignore the second order differentials.

$$\text{Then } \Delta V = \pi (R^2 \Delta d + 2R \Delta R d)$$

$$\frac{\Delta V}{V} = \frac{\pi R^2 \Delta d}{\pi R^2 d} + \frac{2\pi R \Delta R d}{\pi R^2 d} = \frac{\Delta d}{d} + 2 \cdot \frac{\Delta R}{R} = 3 \frac{\Delta D}{D} \quad (\because \frac{\Delta d}{d} = \frac{\Delta D}{D})$$

(4)

42E. The heat needed is

$$\begin{aligned}
 Q &= \rho V L_F = m L_F \\
 &= (0.1) \cdot (1 \times 10^3 \text{ kg/m}^3) \cdot (333 \text{ kJ/kg}) \\
 &= 6.7 \times 10^{12} \text{ J}
 \end{aligned}$$

82E. (a) The rate of heat flow is

$$\begin{aligned}
 H &= \frac{KA(T_H - T_C)}{L} \\
 &= \frac{(0.040 \text{ W/mK}) \cdot (1.8 \text{ m}^2) \cdot (33^\circ\text{C} - 1.0^\circ\text{C})}{1.0 \times 10^{-2} \text{ m}} \\
 &= 2.3 \times 10^2 \text{ J/s}
 \end{aligned}$$

(b) The new rate of heat flow is

$$H' = \frac{K'H}{K} = \frac{(0.60 \text{ W/mK})(230 \text{ J/s})}{0.040 \text{ W/mK}} = 3.5 \times 10^3 \text{ J/s}$$

$$\left(\because \frac{H'}{K'} = \frac{H}{K} = \frac{A(T_H - T_C)}{L} \right)$$

which is about 15 times as fast as the original heat flow.

(5)

92P. (a) surface area of the sphere = $4\pi r^2$

$$\begin{aligned} P_r &= \sigma \epsilon A T^4 \\ &= (5.6703 \times 10^{-8} \text{ W/m}^2 \text{K}^4)(0.850) \cdot (4\pi)(0.5 \text{ m}^2)(300 \text{ K})^4 \\ &= 1.23 \times 10^3 \text{ W} \end{aligned}$$

$$(b) P_a = \sigma \epsilon A T_{\text{environment}}^4$$

$$\begin{aligned} &= (5.6703 \times 10^{-8} \text{ W/m}^2 \text{K}^4)(0.850)(4\pi)(0.5 \text{ m}^2)(350 \text{ K})^4 \\ &= 2.27 \times 10^3 \text{ W} \end{aligned}$$

$$\begin{aligned} (c) P_n &= P_a - P_r = 2.27 \times 10^3 \text{ W} - 1.23 \times 10^3 \text{ W} \\ &= 1.04 \times 10^3 \text{ W} \end{aligned}$$